

O-Alkylation of 3-Pyridinols

POUL NEDENSKOV, NIELS CLAUSON-KAAS, JØRGEN LEI, HENNING HEIDE, GERT OLSEN and GERT JANSEN

28 Rugmarken, DK-3520 Farum, Denmark

A general recipe for the preparation of 3-alkoxy-pyridines by alkylation of 3-pyridinols in dimethyl sulfoxide is given. The recipe has been used to prepare 35 new 3-alkoxy-pyridines. Alkylation of disodium salts of 3-hydroxy-2-pyridones under the same conditions gave 1-alkyl-3-alkoxy-2-pyridones. Of this new group of compounds three have been prepared.

Most alkylations of 3-pyridinols reported in the literature give low yields of 3-alkoxy-pyridines. Alkylation preferably takes place at the pyridine nitrogen atom, whereby quaternary pyridinium compounds are formed (*cf.* Refs. 1, 2).

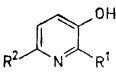
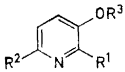
We have found (*cf.* also Ref. 3) that sodium salts of 3-pyridinols are preferentially *O*-alkylated in dimethyl sulfoxide (DMSO), and have worked out a general recipe for such reactions. The recipe is simple and usually gives good yields of 3-alkoxy-pyridines. A solution of the sodium salt is first prepared by dissolving the 3-pyridinol in a methanolic solution of an equimolar amount of sodium methoxide. Then DMSO is added and all methanol removed by distillation. Finally the alkyl halide is added to the resulting solution or suspension of the sodium salt, whereby alkylation takes place.

The recipe has been used to prepare 35 new 3-alkoxy-pyridines (Table 1). Their structures follow from the syntheses, from analyses, and from the fact that the compounds gave no color reaction with a ferric chloride solution.

Alkylation of the disodium salts of 3-hydroxy-2-pyridones after the recipe was also attempted. It was found that *N,O*-dialkylation takes place. The reactions carried out are shown in Table 2. The three reaction products are new. They give no color reaction with a ferric chloride solution. The IR spectra showed absorption at 1645—1660 cm^{-1} (amide-carbonyl). These facts, together with the syntheses and the analyses, prove the structures shown in Table 2. No 1-alkyl-3-alkoxy-2-pyridones are known from the literature.

Preliminary experiments showed that dimethylformamide is as good a solvent for the preparation of 3-alkoxy-pyridines as is DMSO. When we have used DMSO throughout, it is because only this solvent enables one to prepare sodium salts of 3-pyridinols *in situ* in the easy way described here.

Table 1. 3-Alkoxy pyridines.

3-Pyridinol		Alkylating agent XR^3	3-Alkoxy pyridine		
				No.	Yield %
R ¹	R ²	X	R ³		
H	H	Cl	$-(\text{CH}_2)_1\text{CH}_3$	I	50
H	H	Cl	$-\text{CH}_2\text{CON}(\text{C}_2\text{H}_5)_2$	II	62
Br	H	I	$-\text{CH}_3$	III	60
Br	H	Cl	$-\text{CH}_2\text{COOCH}_3$	IV	82
Br	H	Cl	$-\text{CH}_2\text{COC}_6\text{H}_5$	V	45
Br	H	Cl	A	VI	44
Cl	H	Cl	A	VII	35
Cl	H	Cl	B	VIII	74
NO_2	H	Cl	A	IX	39
H	CH_3	Cl	$-\text{CH}_2\text{COOCH}_3$	X	58
H	CH_3	Cl	$-\text{CH}_2\text{COOC}_2\text{H}_5$	XI	86
H	CH_3	Cl	$-\text{CH}_2\text{CONH}_2$	XII	52
NO_2	CH_3	I	$-\text{CH}_3$	XIII	77
NH_2	H	Br	$-\text{CH}_2\text{CH}=\text{CH}_2$	XIV	34
NH_2	H	Br	$-\text{CH}_2\text{C}\equiv\text{CH}$	XV	45
NH_2	H	Br	$-\text{CH}_2\text{CH}(\text{OCH}_3)_2$	XVI	59
NH_2	H	Cl	$-(\text{CH}_2)_3\text{N}(\text{CH}_3)_2$	XVII	58
NH_2	H	Cl	$-\text{CH}_2\text{C}_6\text{H}_5$	XVIII	61
CH_2OH	H	Cl	$-\text{CH}_2\text{CONH}_2$	XIX	47
COOCH_3	H	I	$-\text{CH}_3$	XX	59
COOCH_3	H	Cl	$-\text{CH}_2\text{COOCH}_3$	XXI	81
COOCH_3	H	Cl	$-\text{CH}_2\text{CONH}_2$	XXII	40
CONH_2	H	I	$-\text{CH}_3$	XXIII	40

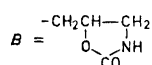
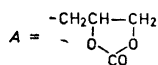


Table 1. Continued.

CONH ₂	H	Br	-(CH ₂) ₃ CH ₃	XXIV	52
—	—	Br	-CH ₂ CH=CH ₂	XXV	64
—	—	Br	-CH ₂ CH(CH ₃) ₂	XXVI	52
—	—	Cl	-CH ₂ C ₆ H ₅	XXVII	66
—	—	Cl	-CH ₂ COOCH ₃	XXVIII	63
—	—	Cl	-CH ₂ CON(C ₂ H ₅) ₂	XXIX	37
—	—	Br	-CHCOOC ₂ H ₅ C ₂ H ₅	XXX	45
CN	H	I	-CH ₃	XXXI	29
—	—	Br	-CH ₂ C ₆ H ₅	XXXII	45
C	H	Cl	-CH ₂ COOCH ₃	XXXIII	35
D	H	Br	-CH ₂ CH=CH ₂	XXXIV	41
E	H	Br	-CH ₂ CH=CH ₂	XXXV	60

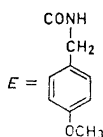
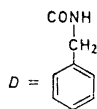
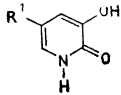
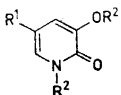


Table 2. 1-Alkyl-3-alkoxy-2(1H)-pyridones.

3-Hydroxy-2-(1H)-pyridone	Alkylating agent	Reaction product	
	Hal R ²		
		No.	Yield %
R ¹ = H	ClCH ₂ C ₆ H ₅	XXXVI	80
R ¹ = H	ClCH ₂ COOCH ₃	XXXVII	42
R ¹ = Cl	Br(CH ₂) ₃ CH ₃	XXXVIII	41

EXPERIMENTAL

O-Alkylation of 3-pyridinols, general recipe. The 3-pyridinol (0.100 mole) is added to a solution of sodium methoxide in methanol [from sodium (2.30 g, 0.100 mole) and methanol (25 ml)] in a 250 ml reaction flask. So much DMSO, as is necessary to get a clear solution, is added (usually 100–150 ml). The mixture may be heated to 80–90° to accelerate dissolution. The solution is evaporated on the water bath under reduced pressure, until pure DMSO distills (b.p.₁₀ 69°). The remaining methanol-free solution or suspension of sodium salt is cooled to room temperature. The alkylating agent (0.100 mole) is added in one portion with vigorous stirring. The reaction temperature is kept below 30° and strong cooling may be necessary the first few minutes. Fast reactions are complete after 15 min or less and the reaction mixture may be worked up immediately. Slow reactions are left standing with slow stirring at room temperature overnight. In 5 cases, mentioned particularly in the following list of preparations, it was necessary to keep the reaction mixture at 80° for a number of hours. The resulting reaction mixture, which is usually turbid or contains varying amounts of crystals, is evaporated to dryness on the water bath (100°) under 10 mm. Water is added to the residue and the alkoxy-pyridine isolated by filtration, or, if no precipitate is formed, by extraction with chloroform. Unreacted 3-pyridinol may be removed from the chloroform extract by shaking with N sodium hydroxide solution. The crude product is usually easy to purify, by distillation, or by crystallization. The 35 alkoxy-pyridines prepared are listed below. They gave no color reaction with a ferric chloride solution.

3-Pentyloxy-pyridine (I). From 3-pyridinol and 1-chloropentane; b.p.₉ 116°, n_D^{25} 1.4946. [Found: C 72.9; H 9.2; N 8.6. Calc. for C₁₀H₁₅NO (165.2): C 72.7; H 9.2; N 8.5].

N,N-Diethyl-2-(3-pyridyloxy)acetamide (II). From 3-pyridinol and 2-chloro-N,N-diethylacetamide; b.p._{0.4} 137°, n_D^{25} 1.5267. [Found: C 63.7; H 7.9; N 13.5. Calc. for C₁₁H₁₆N₂O₂ (208.3): C 63.4; H 7.7; N 13.5]. A monohydrochloride was prepared in the usual way and crystallized from 2-propanol, yield 69%, m.p. 140–142°. [Found: C 53.7; H 7.2; Cl⁻ 14.7; N 11.5. Calc. for C₁₁H₁₇ClN₂O₂ (244.7): C 54.0; H 7.0; Cl⁻ 14.7; N 11.4].

2-Bromo-3-methoxy-pyridine (III). From 2-bromo-3-pyridinol and iodomethane; b.p.₉ 123°, m.p. 45° [from benzene (40–65°)]. [Found: C 38.3; H 3.1; Br 42.9; N 7.3. Calc. for C₆H₆BrNO (188.0): C 38.5; H 3.2; Br 42.5; N 7.5].

[(2-Bromo-3-pyridyl)oxy]acetic acid methyl ester (IV). From 2-bromo-3-pyridinol and chloroacetic acid methyl ester; m.p. 65–69° (from ether). [Found: C 39.1; H 3.4; Br 32.2; N 5.6; OCH₃ 12.7. Calc. for C₈H₈BrNO (214.1): C 39.0; H 3.3; Br 32.4; N 5.7; one OCH₃ 12.6].

2-[(2-Bromo-3-pyridyl)oxy]acetophenone (V). From 2-bromo-3-pyridinol and 2-chloroacetophenone; m.p. 105–107° (from benzene-cyclohexane 1:1). [Found: C 53.9; H 3.6; Br 27.3; N 4.7. Calc. for C₁₃H₁₀BrNO₂ (292.1): C 53.5; H 3.5; Br 27.4; N 4.8].

3-[(2-Bromo-3-pyridyl)oxy]-1,2-propanediol cyclic carbonate (VI). From 2-bromo-3-pyridinol and 3-chloro-1,2-propanediol cyclic carbonate. The reaction mixture in DMSO was kept at 80° for 40 h, otherwise the directions of the general recipe were followed; m.p. 143–144° (from ethanol). [Found: C 39.4; H 2.9; Br 29.3; N 5.3. Calc. for C₉H₈BrNO₄ (274.1): C 39.4; H 2.9; Br 29.2; N 5.1].

3-[(2-Chloro-3-pyridyl)oxy]-1,2-propanediol cyclic carbonate (VII). From 2-chloro-3-pyridinol and 3-chloro-1,2-propanediol cyclic carbonate as described for the preparation of compound VI; m.p. 124–126° (from dioxane-2-propanol 1:1). [Found: C 46.8; H 3.6; Cl 15.4; N 6.0. Calc. for C₉H₈ClNO₄ (229.6): C 47.1; H 3.5; Cl 15.5; N 6.1].

5-[(2-Chloro-3-pyridyl)oxymethyl]-2-oxazolidinone (VIII). From 2-chloro-3-pyridinol and 5-chloromethyl-2-oxazolidinone as described for the preparation of compound VI; m.p. 186–188° (from dioxane). [Found: C 47.5; H 4.2; Cl 15.3; N 12.4. Calc. for C₉H₈ClN₂O₃ (228.6): C 47.3; H 4.0; Cl 15.5; N 12.3].

3-[(2-Nitro-3-pyridyl)oxy]-1,2-propanediol cyclic carbonate (IX). From 2-nitro-3-pyridinol and 3-chloro-1,2-propanediol cyclic carbonate as described for the preparation of compound VI; m.p. 162–164° (from dioxane). [Found: C 45.3; H 3.5; N 11.7. Calc. for C₉H₈N₂O₆ (240.2): C 45.0; H 3.4; N 11.7].

[(6-Methyl-3-pyridyl)oxy]acetic acid methyl ester (X). From 6-methyl-3-pyridinol and chloroacetic acid methyl ester; m.p. 50–51° [(from ether-benzene (b.p. 40–60°) 1:1].

[Found: C 59.7; H 6.0; N 7.8; OCH₃ 17.2. Calc. for C₉H₁₁NO₃ (181.2): C 59.7; H 6.1; N 7.7; one OCH₃ 17.1].

[(6-Methyl-3-pyridyl)oxy]acetic acid ethyl ester (XI). From 6-methyl-3-pyridinol and chloroacetic acid ethyl ester; b.p._{0.1} 111°, *n*_D²⁵ 1.5029, m.p. 10°. [Found: C 61.8; H 6.8; N 7.3; OC₂H₅ 22.5. Calc. for C₁₀H₁₃NO₃ (195.2): C 61.5; H 6.7; N 7.2; one OC₂H₅ 23.0].

2-[(6-Methyl-3-pyridyl)oxy]acetamide (XII). From 6-methyl-3-pyridinol and 2-chloroacetamide; m.p. 154–156° (from water, dried at 100° under 1 mm). [Found: C 58.0; H 6.2; N 17.0. Calc. for C₉H₁₀N₂O₂ (166.2): C 57.8; H 6.1; N 16.9].

3-Methoxy-6-methyl-2-nitropyridine (XIII). From 6-methyl-2-nitro-3-pyridinol and iodomethane; m.p. 88–89° (from cyclohexane). [Found: C 50.2; H 5.0; N 16.8. Calc. for C₇H₉N₂O₃ (168.2): C 50.0; H 4.8; N 16.7].

3-Allyloxy-2-aminopyridine (XIV). From 2-amino-3-pyridinol and 3-bromopropene; b.p._{0.1} 82–83°, m.p. 42–44° [from ether-benzene (40–65°) 1:2]. [Found: C 63.8; H 6.7; N 18.5. Calc. for C₈H₁₀N₂O (150.2): C 64.0; H 6.7; N 18.7].

2-Amino-3-(2-propynyloxy)pyridine (XV). From 2-amino-3-pyridinol and 3-bromopropene; b.p._{0.2} 91–92°, m.p. 58–59° [from ether-benzene (40–65°) 1:1]. [Found: C 65.1; H 5.8; N 18.9. Calc. for C₈H₈N₂O (148.2): C 64.9; H 5.4; N 18.9].

[(2-Amino-3-pyridyl)oxy]acetaldehyde dimethyl acetal (XVI). From 2-amino-3-pyridinol and bromoacetaldehyde dimethyl acetal. The reaction mixture in DMSO was kept at 80° for 4 h, otherwise the directions of the general recipe were followed; b.p._{0.3} 111°, *n*_D²⁵ 1.5360. [Found: C 54.8; H 7.3; N 14.2; OCH₃ 30.8. Calc. for C₉H₁₄N₂O₃ (198.2): C 54.5; H 7.1; N 14.1; two OCH₃ 31.3].

2-Amino-3-[3-(dimethylamino)propyloxy]pyridine (XVII). From 2-amino-3-pyridinol and 3-chloro-*N,N*-dimethylpropylamine; b.p._{0.2} 112°, m.p. 67–69° (from cyclohexane). [Found: C 61.3; H 8.9; N 21.5. Calc. for C₁₀H₁₇N₃O (195.3): C 61.5; H 8.8; N 21.5].

2-Amino-3-benzoyloxy pyridine (XVIII). From 2-amino-3-pyridinol and α-chloro-toluene; b.p._{0.05} 150°, m.p. 96–97° (from methanol). [Found: C 72.2; H 6.2; N 14.0. Calc. for C₁₂H₁₂N₂O (200.2): C 72.0; H 6.0; N 14.0].

2-[(2-Hydroxymethyl-3-pyridyl)oxy]acetamide (XIX). From 3-hydroxy-2-pyridine-methanol and 2-chloroacetamide; m.p. 157–158° (from 2-propanol). [Found: C 52.6; H 5.7; N 15.6. Calc. for C₈H₁₀N₂O₃ (182.2): C 52.7; H 5.5; N 15.4].

3-Methoxypicolinic acid methyl ester (XX). From 3-hydroxypicolinic acid methyl ester and iodomethane; b.p._{0.1} 85°, *n*_D²⁵ 1.5240. [Found: C 57.5; H 5.4; N 8.3; OCH₃ 37.3. Calc. for C₈H₉NO₂ (167.2): C 57.5; H 5.4; N 8.4; two OCH₃ 37.1].

3-(Methoxycarbonylmethoxy)picolinic acid methyl ester (XXI). From 3-hydroxypicolinic acid methyl ester and chloroacetic acid methyl ester; m.p. 78–79° (from benzene). [Found: C 53.6; H 4.6; N 6.3. Calc. for C₁₀H₁₁NO₅ (225.2): C 53.3; H 4.9; N 6.2].

3-(Carbamoylmethoxy)picolinic acid methyl ester (XXII). From 3-hydroxypicolinic acid methyl ester and 2-chloroacetamide; m.p. 185–186° (from 2-propanol). [Found: C 51.0; H 4.6; N 13.5; OCH₃ 14.5. Calc. for C₉H₁₀N₂O₄ (210.2): C 51.4; H 4.8; N 13.3; one OCH₃ 14.7].

3-Methoxypicolinamide (XXIII). From 3-hydroxypicolinamide and iodomethane; m.p. 198–200° (from propanol). [Found: C 55.3; H 5.8; N 18.5; OCH₃ 19.6. Calc. for C₇H₈N₂O₂ (152.2): C 55.3; H 5.3; N 18.4; one OCH₃ 20.4].

3-Butoxypicolinamide (XXIV). From 3-hydroxypicolinamide and 1-bromobutane; m.p. 109–110° (from toluene). [Found: C 61.9; H 7.3; N 14.7. Calc. for C₁₀H₁₄N₂O₂ (194.2): C 61.8; H 7.3; N 14.4].

3-Allyloxypicolinamide (XXV). From 3-hydroxypicolinamide and 3-bromopropene; m.p. 125–126° (from toluene). [Found: C 60.5; H 6.0; N 15.4. Calc. for C₉H₁₀N₂O₂ (178.2): C 60.7; H 5.7; N 15.7].

3-Isobutoxypicolinamide (XXVI). From 3-hydroxypicolinamide and 1-bromo-2-methylpropane; m.p. 109–110° (from toluene). [Found: C 62.0; H 7.4; N 14.6. Calc. for C₁₀H₁₄N₂O₂ (194.1): C 61.8; H 7.3; N 14.4].

3-Benzoyloxypicolinamide (XXVII). From 3-hydroxypicolinamide and α-chloro-toluene; m.p. 106–107° (from toluene). A polymorphic form, m.p. 116–117°, is obtained by sublimation (130°, 0.1 mm). [Found, low melting form: C 68.4; H 5.4; N 12.4. Found, high melting form: C 68.7; H 5.2; N 12.5. Calc. for C₁₃H₁₂N₂O₂ (228.2): C 68.4; H 5.3; N 12.3].

[(2-Carbamoyl-3-pyridyl)oxy]acetic acid methyl ester (XXVIII). From 3-hydroxypicolinamide and chloroacetic acid methyl ester; m.p. 141–143° (from methanol-2-propanol 1:4). [Found: C 51.7; H 4.9; N 13.5; OMe 14.9. Calc. for C₉H₁₀N₂O₄ (210.2): C 51.4; H 4.8; N 13.3; one OCH₃ 14.7].

3-[(N,N-Diethylcarbamoyl)methoxy]picolinamide (XXIX). From 3-hydroxypicolinamide and 2-chloro-N,N-diethylacetamide; m.p. 134–135°. [Found: C 57.4; H 6.8; N 16.7. Calc. for C₁₂H₁₇N₃O₃ (251.3): C 57.4; H 6.8; N 16.7].

2-[(2-Carbamoyl-3-pyridyl)oxy]butyric acid ethyl ester (XXX). From 3-hydroxypicolinamide and 2-bromobutyric acid ethyl ester; m.p. 101–108° (from toluene). [Found: C 57.3; H 6.4; N 11.0; OC₂H₅ 17.8. Calc. for C₁₂H₁₆N₂O₄ (252.3): C 57.1; H 6.4; N 11.1; one OC₂H/ 17.9]. Sublimation as well as repeated crystallizations do not change the analytical values nor the unusually large melting range.

3-Methoxypicolinonitrile (XXXI). From 3-hydroxypicolinonitrile and iodomethane; m.p. 111–113° (from 2-propanol). [Found: C 62.6; H 4.7; N 21.0. Calc. for C₇H₆N₂O (134.1): C 62.7; H 4.5; N 20.9].

3-(Benzoyloxy)picolinonitrile (XXXII). From 3-hydroxypicolinonitrile and α-chlorotoluene; m.p. 99–101° (from 2-propanol). [Found: C 74.5; H 5.0; N 13.2. Calc. for C₁₃H₁₀N₂O (210.2): C 74.3; H 4.8; N 13.3].

{[2-(N-Phenylcarbamoyl)-3-pyridyl]oxy}acetic acid methyl ester (XXXIII). From 3-hydroxy-N-phenylpicolinamide and chloroacetic acid methyl ester; m.p. 109–110° (from toluene). [Found: C 63.1; H 5.1; N 10.1. Calc. for C₁₅H₁₄N₂O₄ (286.3): C 62.9; H 4.9; N 9.8].

3-Allyloxy-N-benzylpicolinamide (XXXIV). From 3-hydroxy-N-benzylpicolinamide and 3-bromopropene; m.p. 72–73° (from 2-propanol). [Found: C 71.8; H 6.1; N 10.3. Calc. for C₁₈H₁₆N₂O₂ (268.3): C 71.6; H 6.0; N 10.4].

3-Allyloxy-N-(p-methoxybenzyl)picolinamide (XXXV). From 3-hydroxy-N-(p-methoxybenzyl)picolinamide and 3-bromopropene; m.p. 74–75° (from 2-propanol). [Found: C 68.2; H 6.1; N 9.6. Calc. for C₁₇H₁₈N₂O₃ (298.3): C 68.4; H 6.1; N 9.4].

O,N-Alkylations of 3-hydroxy-2(1H)-pyridones. These reactions were carried out after the general recipe by substitution of 0.100 mole of the 3-pyridinol by 0.050 mole of the 3-hydroxy-2(1H)-pyridone. The three products prepared gave no color reaction with a ferric chloride solution.

1-Benzyl-3-benzoyloxy-2(1H)-pyridone (XXXVI). From 3-hydroxy-2(1H)-pyridone and α-chlorotoluene; m.p. 115–116° (from ethanol-water 2:1). IR absorption (in KBr) 1645 cm⁻¹. [Found: C 78.4; H 6.0; N 5.0. Calc. for C₁₉H₁₇NO₂ (291.3): C 78.3; H 5.9; N 4.8].

3-(Methoxycarbonylmethoxy)-2-oxo-1(2H)-pyridineacetic acid methyl ester (XXVII). From 3-hydroxy-2(1H)-pyridone and chloroacetic acid methyl ester; m.p. 102–103° (from 2-propanol-ether 1:1), IR absorption (in KBr) 1660 cm⁻¹. [Found: C 51.4; H 5.3; N 5.6; OCH₃ 23.7. Calc. for C₁₁H₁₃NO₆ (255.2): C 51.8; H 5.1; N 5.5; two OCH₃ 24.3].

3-Butoxy-1-butyl-5-chloro-2(1H)-pyridone (XXXVIII). From 5-chloro-3-hydroxy-2(1H)-pyridone and 1-bromobutane; b.p._{0.3} 139–140°, n_D²⁵ 1.5289, m.p. 46–47° [from ether-benzene (40–65°) 1:1], IR absorption (in KBr) 1660 cm⁻¹. [Found: C 60.6; H 7.7; Cl 13.7; N 5.2. Calc. for C₁₃H₂₀ClNO₂ (257.8): C 60.6; H 7.8; Cl 13.8; N 5.4].

Starting materials. All starting materials used are known compounds, which were obtained in an analytically pure grade from Wolff & Kaaber, Farum, Denmark.

REFERENCES

1. Fürst, H. and Dietz, H. J. *J. prakt. Chem.* [4] **4** (1956) 147.
2. Shapiro, S. L., Weinberg, K. and Freemann, L. *J. Am. Chem. Soc.* **81** (1959) 5140.
3. Mertes, M. P., Borne, R. F. and Hare, L. E. *J. Heterocyclic Chem.* **5** (1968) 281.

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